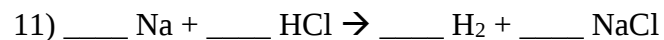
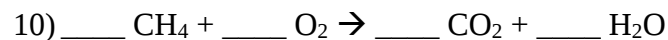
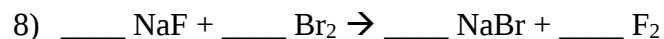
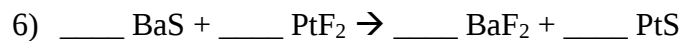
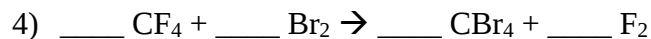
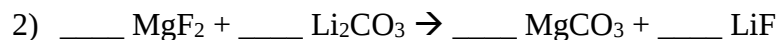
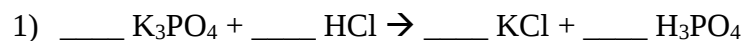


Balancing Chemical Equations – Practice Part 2



12) How many total molecules of product were made in #8? _____

13) How many total molecules of reactants were used in #5? _____

14) How many reactant types are used in #11? _____

15) How many types of products are made in #9? _____

16) 4 molecules of a substance contains 3 atoms of sodium, 1 atom of phosphorus, and 4 atoms of oxygen. What is the correct chemical formula for this substance?